

Assignment

Two Stage Tandem Solar Cell

Objective

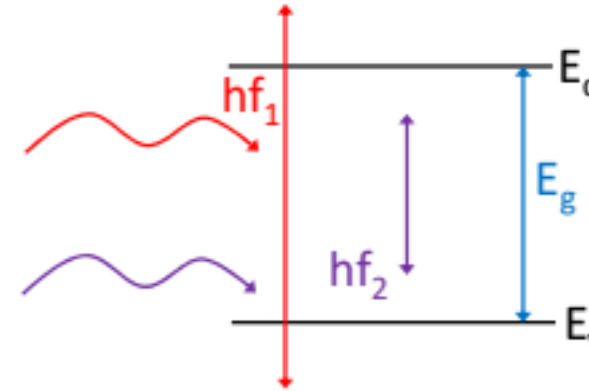
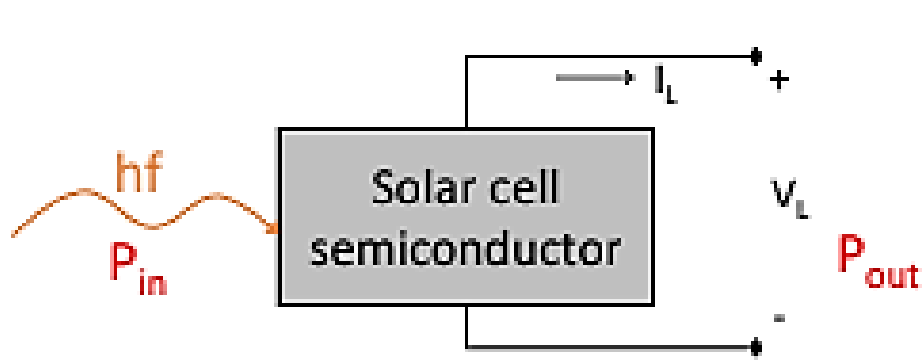
- **Choose two different energy bandgaps of a ternary semiconductor system for application in a two stage tandem solar cell.**
- **Minimize spectral losses.**
- **Justify through necessary calculations and analysis.**

Outcomes

The assignment will address the following course outcomes (CO):

- **CO3:** Compare different materials and select the most appropriate one for specific electrical engineering application
- **CO4:** Demonstrate the capacity to extend learning beyond classroom lectures and activities
- **CO5:** Describe environmental issues in processing and recycling of electronic materials

Theory

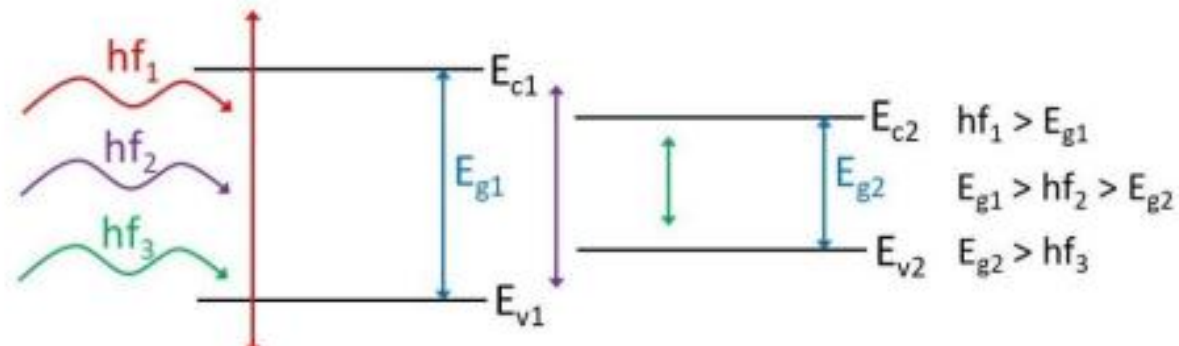
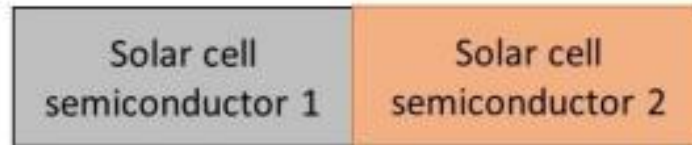


$hf_1 > E_g$

$hf_2 < E_g$

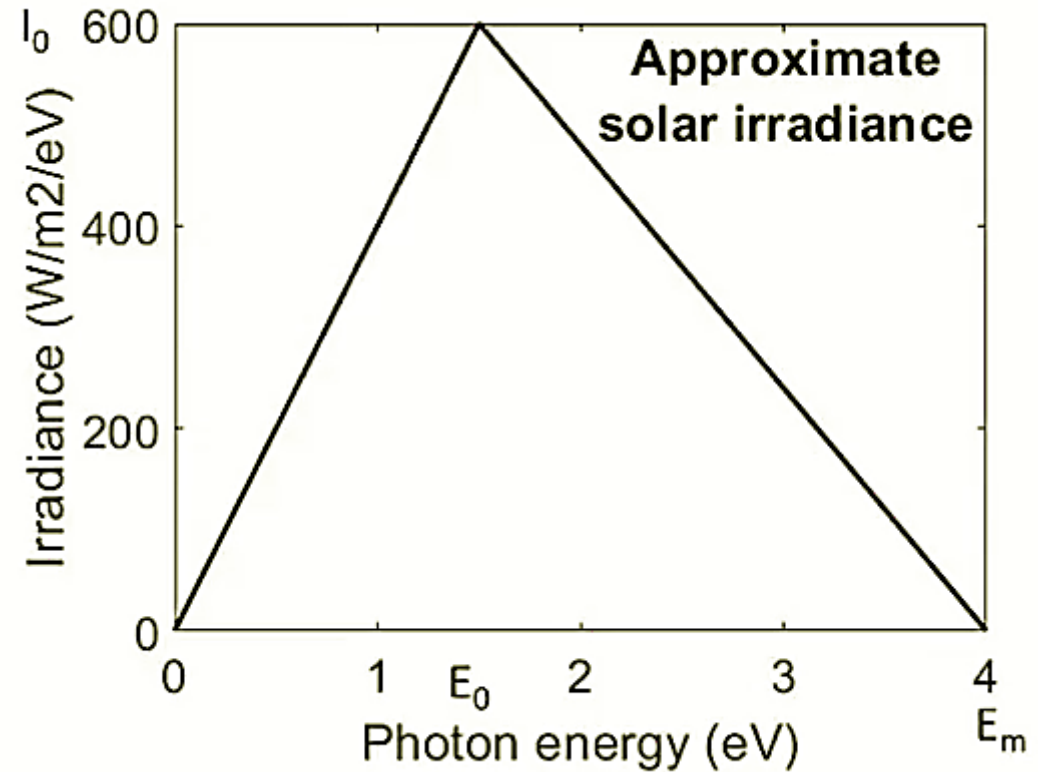
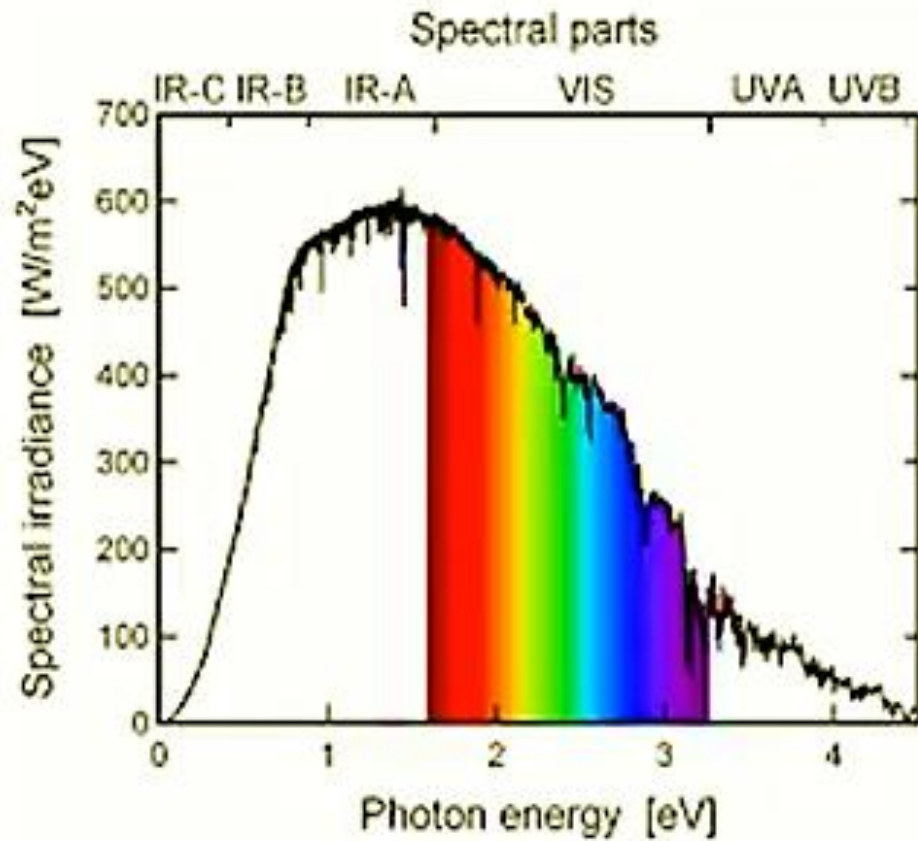
Spectral losses:
 i) $hf_1 - E_g$
 ii) hf_2

Two Stage Tandem Solar Cell

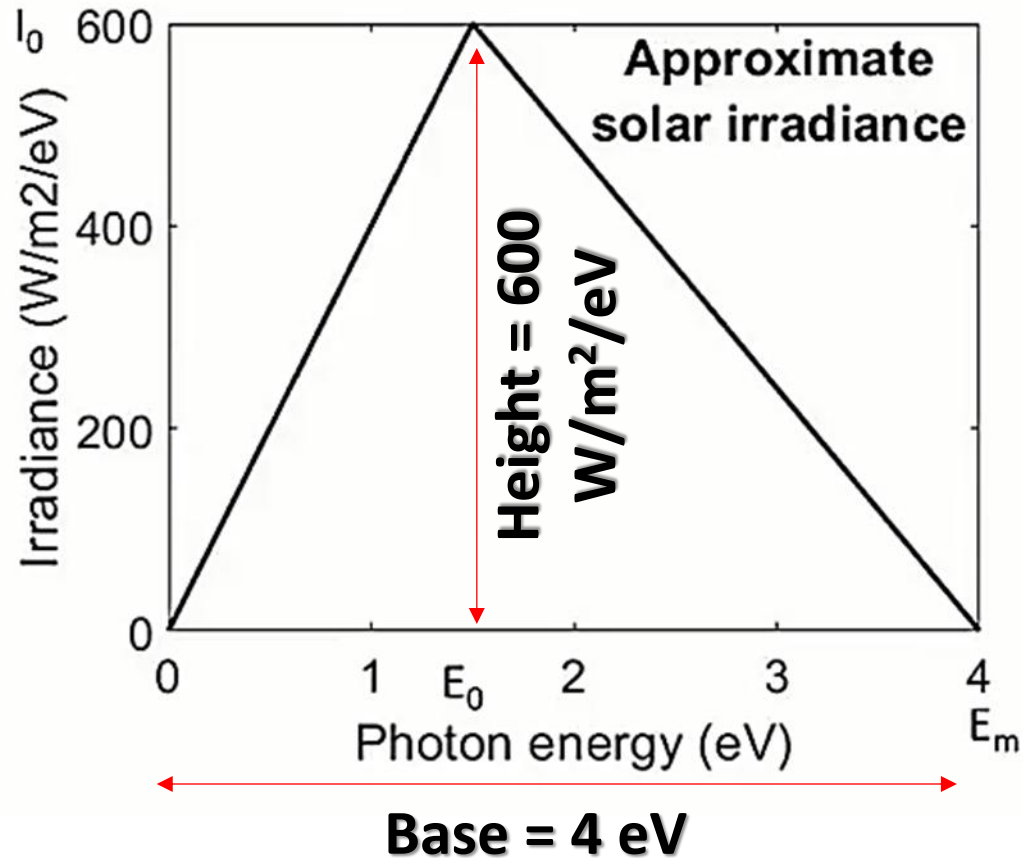


Spectral losses:
 i) $hf_1 - E_{g1}$
 ii) $hf_2 - E_{g2}$
 iii) hf_3

Solar Irradiance Vs Photon Energy



Input Power Calculation



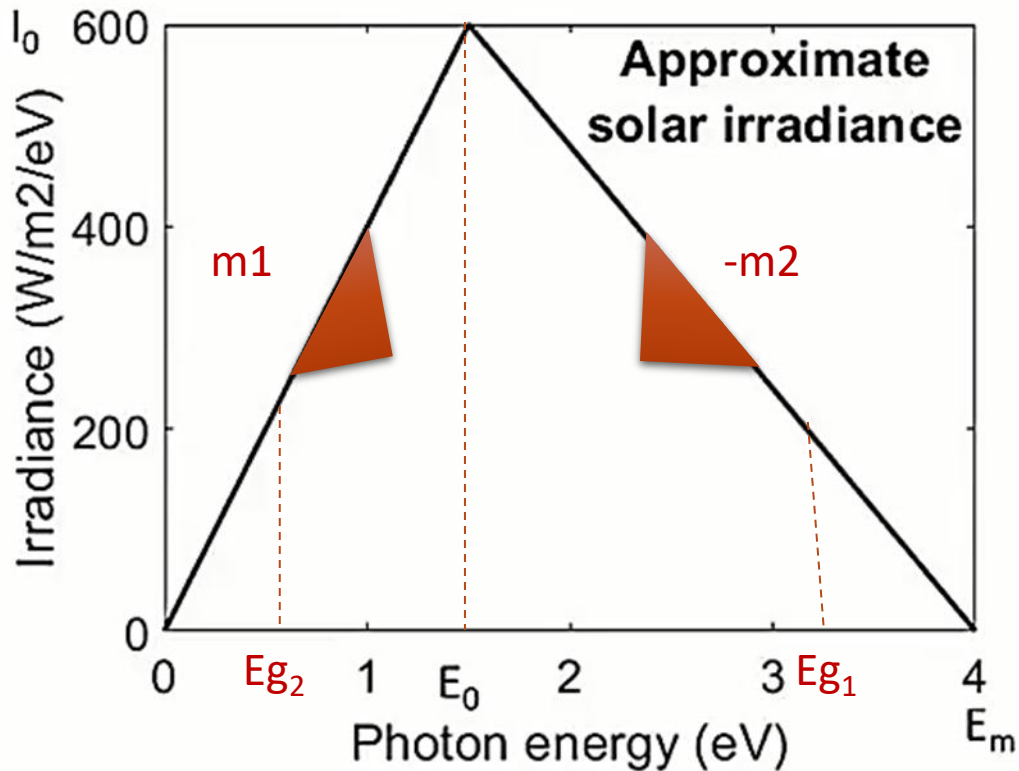
Input Power ,

**$P_{in} = \text{Area of the Triangle}$
(Representing Irradiance Vs Photon Energy)**

$$\begin{aligned} P_{in} &= \frac{1}{2} \times \text{Base} \times \text{Height} \\ &= \frac{1}{2} \times 4\text{eV} \times 600\text{W/m}^2/\text{eV} \\ &= 1200 \text{ W/m}^2 \end{aligned}$$

Output Power Equation

Case 1: $E_{g2} < E_0 < E_{g1}$



$$I(\text{absorbed}) = \begin{cases} I\left(\frac{E_{g1}}{E}\right); & E_{g1} < E_m \\ I\left(\frac{E_{g2}}{E}\right); & E_{g2} < E_0 < E_{g1} \end{cases}$$

$$I = \begin{cases} m1 \times E; & 0 < E < E_0 \\ I_0 - m2(E - E_0); & E_0 < E < E_m \end{cases}$$

$$m1 = I_0/E_0$$

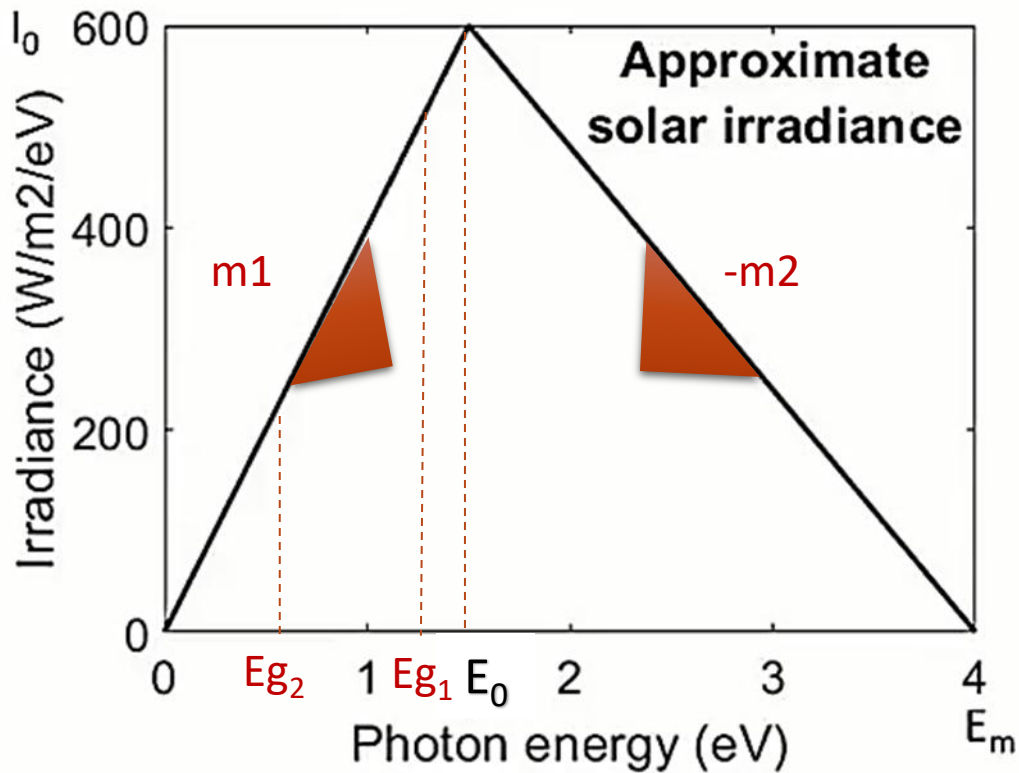
$$m2 = I_0/(E_m - E_0)$$

Case 1

$$\begin{aligned} P_{out} &= \int_{E_{g2}}^{E_m} I(\text{absorbed}) dE = \int_{E_{g2}}^{E_0} I(\text{absorbed}) dE + \int_{E_0}^{E_{g1}} I(\text{absorbed}) dE + \int_{E_{g1}}^{E_m} I(\text{absorbed}) dE \\ &= \int_{E_{g2}}^{E_0} m_1 E \left(\frac{E_{g2}}{E} \right) dE + \int_{E_0}^{E_{g1}} \{I_0 - m_2(E - E_0)\} \left(\frac{E_{g2}}{E} \right) dE + \int_{E_{g1}}^{E_m} \{I_0 - m_2(E - E_0)\} \left(\frac{E_{g1}}{E} \right) dE \\ &= m_1.E_{g2}(E_0 - E_{g2}) + I_0.E_{g2}.\ln(E_{g1}/E_0) - m_2.E_{g2}(E_{g1} - E_0) + m_2.E_0.E_{g2}.\ln(E_{g1}/E_0) \\ &\quad + I_0.E_{g1}.\ln(E_m/E_{g1}) - m_2.E_{g1}(E_m - E_{g1}) + m_2.E_0.E_{g1}.\ln(E_m/E_{g1}) \\ &= m_1.E_{g2}(E_0 - E_{g2}) - m_2.E_{g2}(E_{g1} - E_0) + \{I_0 + (m_2.E_0)\}E_{g2}.\ln(E_{g1}/E_0) \\ &\quad - m_2.E_{g1}(E_m - E_{g1}) + \{I_0 + (m_2.E_0)\}E_{g1}.\ln(E_m/E_{g1}) \end{aligned}$$

Output Power Equation

Case 2: $E_{g2} < E_{g1} < E_0$



$$I(\text{absorbed}) = \begin{cases} I\left(\frac{E_{g1}}{E}\right); & E_{g1} < E_m \\ I\left(\frac{E_{g2}}{E}\right); & E_{g2} < E_{g1} \end{cases}$$

$$I = \begin{cases} m1 \times E; & 0 < E < E_0 \\ I_0 - m2(E - E_0); & E_0 < E < E_m \end{cases}$$

$$m1 = I_0/E_0$$

$$m2 = I_0/(E_m - E_0)$$

Case 2

$$P_{out} = \int_{Eg2}^{Em} I(\text{absorbed}) dE = \int_{Eg2}^{Eg1} I(\text{absorbed}) dE + \int_{Eg1}^{E0} I(\text{absorbed}) dE + \int_{E0}^{Em} I(\text{absorbed}) dE$$

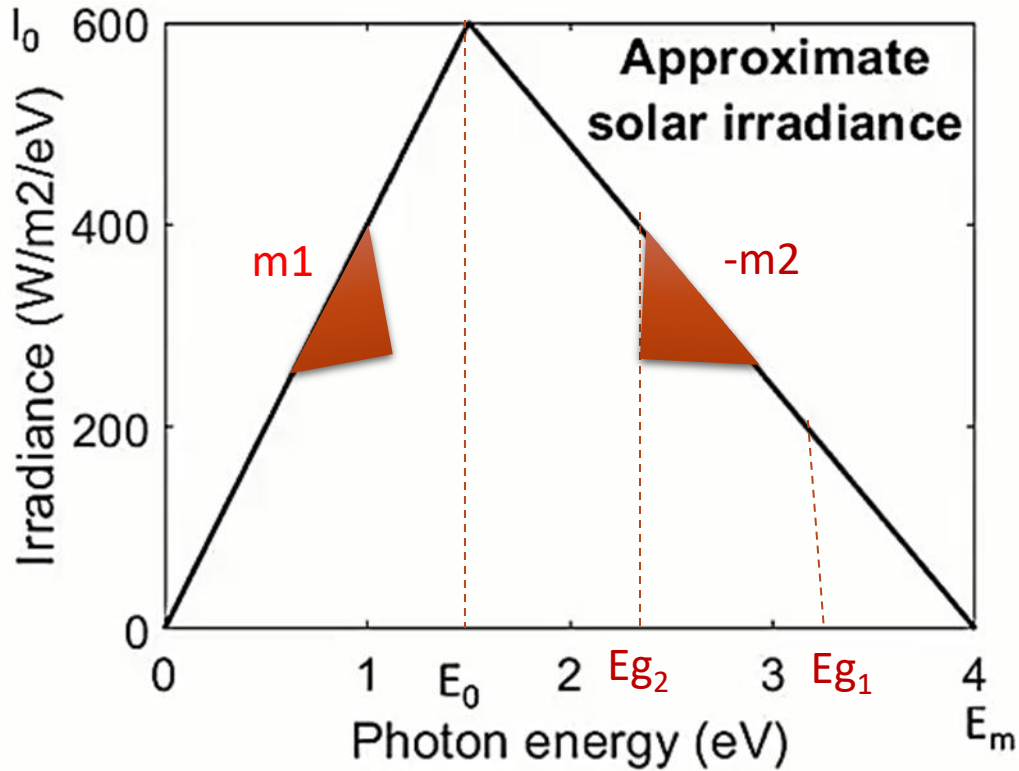
$$= \int_{Eg2}^{Eg1} m_1 E \left(\frac{Eg2}{E} \right) dE + \int_{Eg1}^{E0} m_1 E \left(\frac{Eg1}{E} \right) dE + \int_{E0}^{Em} \{ I_0 - m_2 (E - E0) \} \left(\frac{Eg1}{E} \right) dE$$

$$= m_1.Eg2(Eg1-Eg2) + m_1.Eg1(E0-Eg1) + I_0.Eg1.\ln(Em/E0) - m_2.Eg1(Em-E0) + m_2.E0.Eg1.\ln(Em/E0)$$

$$= m_1.Eg2(Eg1-Eg2) + m_1.Eg1(E0-Eg1) - m_2.Eg1(Em-E0) + \{ I_0 + (m_2.E0) \} Eg1.\ln(Em/E0)$$

Output Power Equation

Case 3: $E_0 < E_{g2} < E_{g1}$



$$I(\text{absorbed}) = \begin{cases} I\left(\frac{E_{g1}}{E}\right); & E_{g1} < E_m \\ I\left(\frac{E_{g2}}{E}\right); & E_{g2} < E_0 < E_{g1} \end{cases}$$

$$I = \begin{cases} m1 \times E; & 0 < E < E_0 \\ I_0 - m2(E - E_0); & E_0 < E < E_m \end{cases}$$

$$m1 = I_0/E_0$$

$$m2 = I_0/(E_m - E_0)$$

Case 3

$$\begin{aligned} P_{out} &= \int_{E_{g2}}^{E_m} I(\text{absorbed}) dE = \int_{E_{g2}}^{E_{g1}} I(\text{absorbed}) dE + \int_{E_{g1}}^{E_m} I(\text{absorbed}) dE \\ &= \int_{E_{g2}}^{E_{g1}} \{I_0 - m^2(E - E_0)\} \left(\frac{E_{g2}}{E}\right) dE + \int_{E_{g1}}^{E_m} \{I_0 - m^2(E - E_0)\} \left(\frac{E_{g1}}{E}\right) dE \\ &= I_0.E_{g2}.\ln(E_{g1}/E_{g2}) - m^2.E_{g2}(E_{g1}-E_{g2}) + m^2.E_0.E_{g2}.\ln(E_{g1}/E_{g2}) \\ &\quad + I_0.E_{g1}.\ln(E_m/E_{g1}) - m^2.E_{g1}(E_m-E_{g1}) + m^2.E_0.E_{g1}.\ln(E_m/E_{g1}) \\ &= \{I_0+(m^2.E_0)\}E_{g2}.\ln(E_{g1}/E_{g2}) - m^2.E_{g2}(E_{g1}-E_{g2}) - m^2.E_{g1}(E_m-E_{g1}) \\ &\quad + \{I_0+(m^2.E_0)\}E_{g1}.\ln(E_m/E_{g1}) \end{aligned}$$



Gallium Arsenide Antimonide

Equation for Bandgap,

$$E_g = 1.42 - 1.9x + 1.2x^2 ; [0 < x < 0.3]$$

$$@ x=0, E_g = (1.42 - 1.9*0 + 1.2*0^2) = 1.42 \text{ eV}$$

$$@ x=0.3, E_g = (1.42 - 1.9*0.3 + 1.2*0.3^2) = 0.958 \text{ eV}$$

So, the bandgap for GaAs_{1-x}Sb_x can be stated as,

$$0.958 < E_g < 1.42$$

GaAs_{1-x}Sb_x

Gallium Arsenide Antimonide

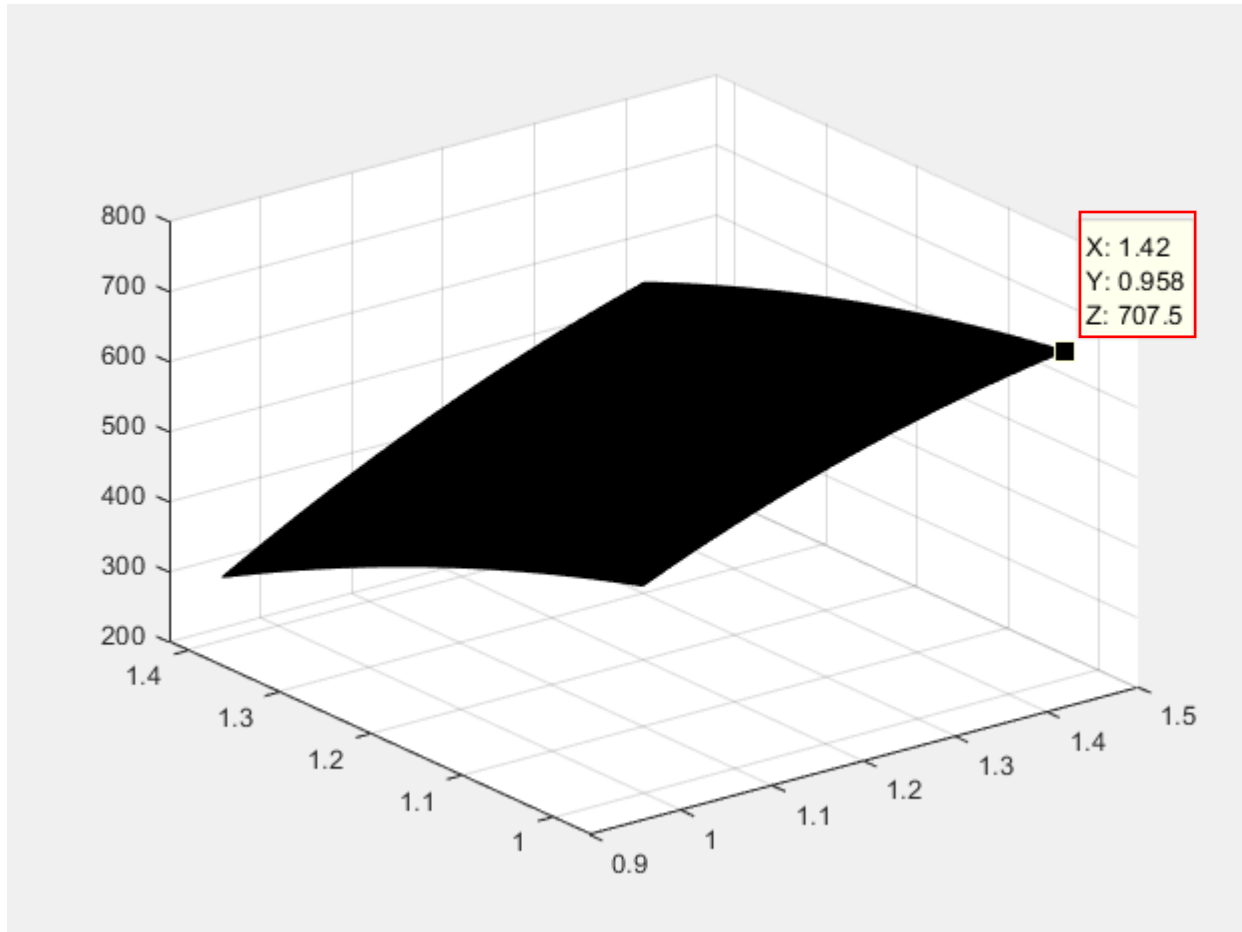
```
1  %%gallium arsenide antimonide
2  - close all
3  - clc
4  - IO = 600;
5  - EO = 1.5;
6  - Em = 4;
7
8  - m1 = IO/EO;
9  - m2 = IO/(Em-EO);
10
11 - Eg1 = 0.958:0.001:1.42;
12 - Eg2 = 0.958:0.001:1.42;
13
14 - Pout=@(Eg1,Eg2) (m1.*Eg2.*(Eg1-Eg2))+ (m1.*Eg1.*(EO-Eg1))...
15 - (m2.*Eg1.*(Em-EO))+((IO+(m2.*EO)).*Eg1.*log(Em./EO)); %%Case 2
16
17 - [EG1, EG2] = meshgrid(Eg1, Eg2);
18 - surf(EG1, EG2, Pout(EG1,EG2));
19 |
20
21
--
```

As the range of the bandgap is lower than E_0 , So we have used the **Case 2** equation to calculate output power P_{out} .

In MATLAB we have used `meshgrid` and `surf` to plot a 3 dimensional figure.



Gallium Arsenide Antimonide



Here,

X axis represents E_{g1} ,

Y axis represents E_{g2} ,

Z axis represents P_{out}

Maximum Value of $P_{out} = 707.5 \text{ W/m}^2$ for

corresponding $E_{g1} = 1.42 \text{ eV}$ and $E_{g2} = 0.958 \text{ eV}$

GaAs_{1-x}Sb_x

Gallium Arsenide Antimonide

Maximum Value of $P_{out} = 707.5 \text{ W/m}^2$
for
corresponding $E_{g1} = 1.42 \text{ eV}$ and $E_{g2} = 0.958 \text{ eV}$

$$E_g = 1.42 - 1.9x + 1.2x^2 ; [0 < x < 0.3]$$

$$@ x=0, E_g = (1.42 - 1.9 \cdot 0 + 1.2 \cdot 0^2) = 1.42 \text{ eV}$$

$$@ x=0.3, E_g = (1.42 - 1.9 \cdot 0.3 + 1.2 \cdot 0.3^2) = 0.958 \text{ eV}$$

Here, $E_{g1} = 1.42 \text{ eV}$ which is the maximum value of E_g
we got from the bandgap equation for corresponding
mol fraction $x_1 = 0$.

$$@ x_1=0, E_{g1}(\text{GaAs}) = 1.42 \text{ eV}$$

Here, $E_{g2} = 0.958 \text{ eV}$ which is the minimum value of E_g
we got from the bandgap equation for corresponding mol
fraction $x_2 = 0.3$.

$$@ x_2=0.3, E_{g2}(\text{GaSb}_{0.3}) = 0.958 \text{ eV}$$



Aluminium Gallium Arsenide

Equation for Bandgap,

$$E_g = \begin{cases} 1.424 + 1.247x & ; x < 0.45 \\ 1.9 + 0.125x + 0.143x^2 & ; x > 0.45 \end{cases}$$

$\text{Al}_x \text{Ga}_{1-x} \text{As}$

Aluminium Gallium Arsenide

For $x < 0.45$,

@ $x=0$,

$$\begin{aligned} E_g &= 1.424 + 1.247x \\ &= 1.424 \text{ eV} \end{aligned}$$

@ $x=0.45$,

$$\begin{aligned} E_g &= 1.424 + 1.247x \\ &= 1.985 \text{ eV} \end{aligned}$$

For $E_g < 1.985 \text{ eV}$

For $x > 0.45$,

@ $x=0.45$,

$$\begin{aligned} E_g &= 1.9 + 0.125x + 0.143x^2 \\ &= 1.985 \text{ eV} \end{aligned}$$

@ $x=1$,

$$\begin{aligned} E_g &= 1.9 + 0.125x + 0.143x^2 \\ &= 2.168 \text{ eV} \end{aligned}$$

For $E_g > 1.985 \text{ eV}$

So, the bandgap for $\text{Al}_x \text{Ga}_{1-x} \text{As}$ can be stated as,

$$1.424 < E_g < 2.168$$



Aluminium Gallium Arsenide

As, the bandgap for $\text{Al}_x \text{Ga}_{1-x}$ is,

$$1.424 < E_g < 2.168$$

So, the output power should be calculated using all three equations.

Case 1 :

$$E_{g2} < E_0 < E_{g1}$$

$$1.50 < E_{g1} < 2.168$$

$$1.424 < E_{g2} < 1.50$$

Case 2 :

$$E_{g2} < E_{g1} < E_0$$

$$1.424 < E_{g1} < 1.50$$

$$1.424 < E_{g2} < 1.50$$

Case 3 :

$$E_0 < E_{g2} < E_{g1}$$

$$1.50 < E_{g1} < 2.168$$

$$1.50 < E_{g2} < 2.168$$

Al_x Ga_{1-x} As

Aluminium Gallium Arsenide

Case 1 :

$$Eg2 < E0 < Eg1$$

$$1.50 < Eg1 < 2.168$$

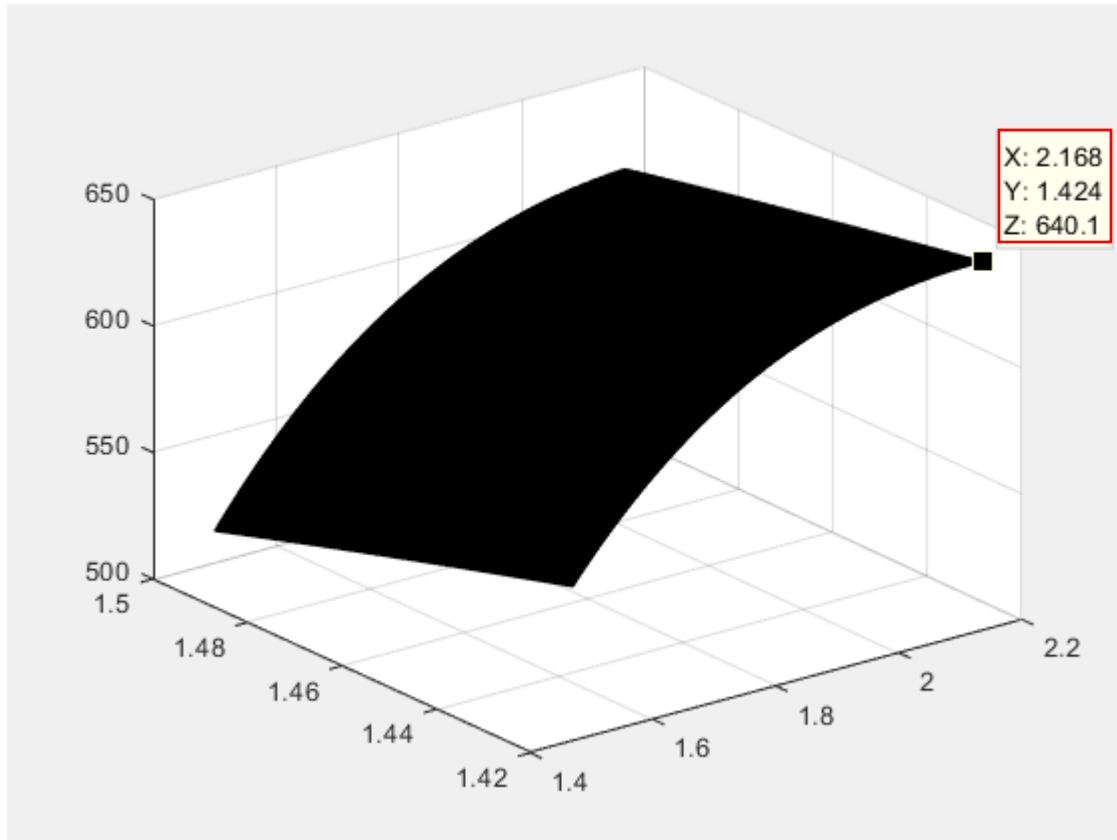
$$1.424 < Eg2 < 1.50$$

```
1 %Aluminium Gallium Arsenide
2 - close all
3 - clc
4
5 - Eg1=1.5:0.001:2.168;
6 - Eg2=1.424:0.001:1.5;
7
8 - IO = 600;
9 - E0 = 1.5;
10 - Em = 4;
11
12 - m1 = IO/E0;
13 - m2 = IO/(Em-E0);
14
15 - Pout = @(Eg1,Eg2) m1.*Eg2.*(E0-Eg2)-m2.*Eg2.*(Eg1-E0)+...
16 (IO+(m2.*E0)).*Eg2.*log(Eg1./E0)- m2.*Eg1.*(Em-Eg1)+...
17 (IO+(m2.*E0)).*Eg1.*log(Em./Eg1); %case 1
18
19 - [EG1, EG2] = meshgrid(Eg1, Eg2);
20 - surf(EG1, EG2, Pout(EG1,EG2));
21
```

In MATLAB we have used meshgrid and surf to plot a 3 dimensional figure.

$\text{Al}_x \text{Ga}_{1-x} \text{As}$

Aluminium Gallium Arsenide



Here,

X axis represents E_{g1} ,
Y axis represents E_{g2} ,
Z axis represents P_{out}

Maximum Value of $P_{out} = 640.1 \text{ W/m}^2$ for

corresponding $E_{g1} = 2.168 \text{ eV}$ and $E_{g2} = 1.424 \text{ eV}$

Al_x Ga_{1-x} As

Aluminium Gallium Arsenide

```
1 %Aluminium Gallium Arsenide
2 - close all
3 - clc
4
5 - Eg1=1.424:0.001:1.5;
6 - Eg2=1.424:0.001:1.5;
7
8 - IO = 600;
9 - EO = 1.5;
10 - Em = 4;
11
12 - m1 = IO/EO;
13 - m2 = IO/(Em-EO);
14
15 - Pout = @(Eg1,Eg2) m1.*Eg2.*(Eg1-Eg2)+ m1.*Eg1.*(EO-Eg1)- ...
16         m2.*Eg1.*(Em-EO)+(IO+m2.*EO).*Eg1.*log(Em/EO);%Case 2
17
18 - [EG1, EG2] = meshgrid(Eg1, Eg2);
19 - surf(EG1, EG2, Pout(EG1,EG2));
```

Case 2:

$$Eg2 < Eg1 < E0$$

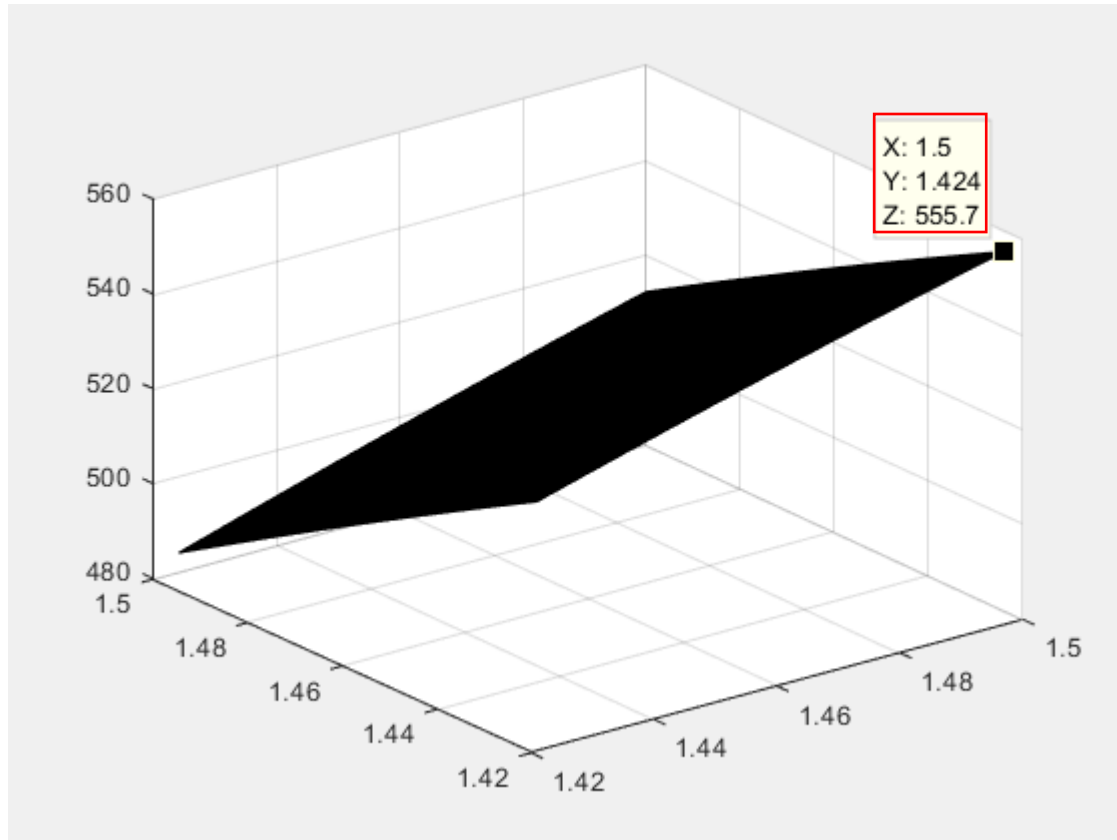
$$1.424 < Eg1 < 1.50$$

$$1.424 < Eg2 < 1.50$$

In MATLAB we have used meshgrid and surf to plot a 3 dimensional figure.



Aluminium Gallium Arsenide



Here,

X axis represents E_{g1} ,

Y axis represents E_{g2} ,

Z axis represents P_{out}

Maximum Value of $P_{out} = 555.7 \text{ W/m}^2$ for

corresponding $E_{g1} = 1.5 \text{ eV}$ and $E_{g2} = 1.424 \text{ eV}$

Al_xGa_{1-x}As

Aluminium Gallium Arsenide

```
1 %Aluminium Gallium Arsenide
2 - close all
3 - clc
4
5 - Eg1=1.5:0.001:2.168;
6 - Eg2=1.5:0.001:2.168;
7
8 - IO = 600;
9 - EO = 1.5;
10 - Em = 4;
11
12 - m1 = IO/EO;
13 - m2 = IO/(Em-EO);
14
15 - Pout = @(Eg1,Eg2) (IO+m2.*EO).*Eg2.*log(Eg1./Eg2) ...
16 -m2.*Eg2.*(Eg1-Eg2)-m2.*Eg1.*(Em-Eg1) ...
17 +(IO+m2.*EO).*Eg1.*log(Em./Eg1); %Case 3
18
19 - [EG1, EG2] = meshgrid(Eg1, Eg2);
20 - surf(EG1, EG2, Pout(EG1,EG2));
21
```

Case 3 :

$E0 < Eg2 < Eg1$

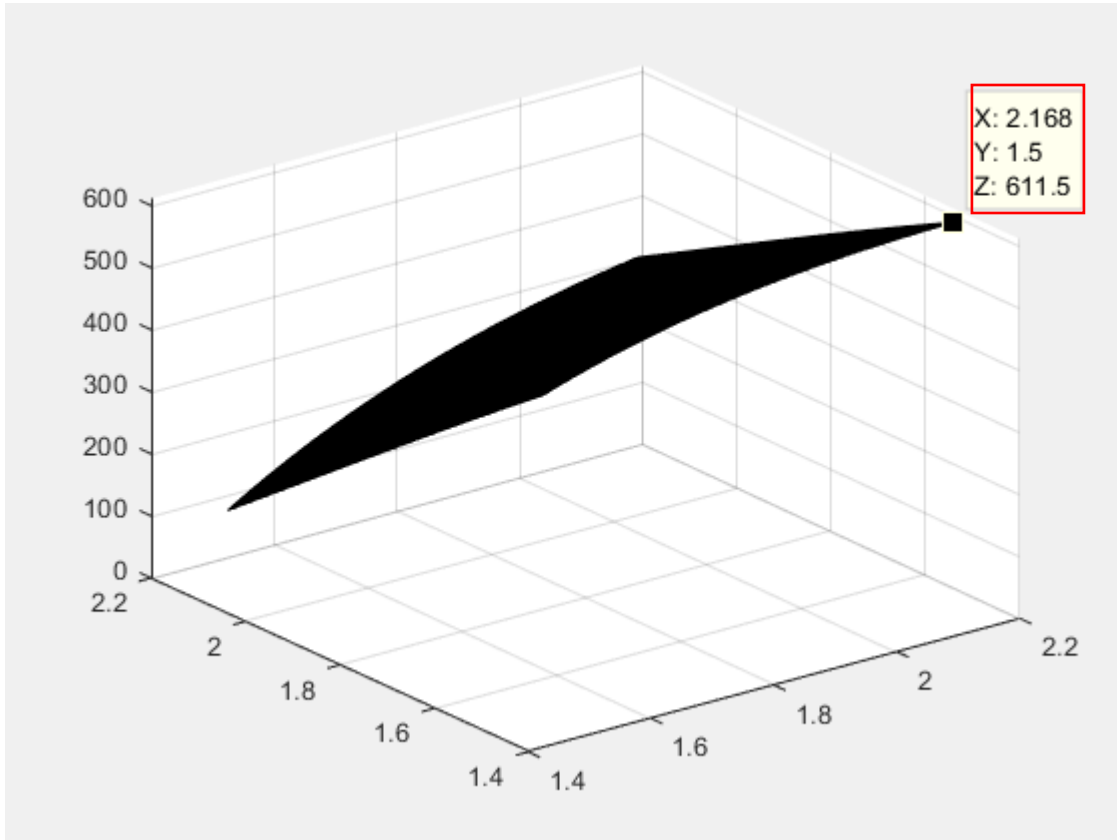
$1.50 < Eg1 < 2.168$

$1.50 < Eg2 < 2.168$

**In MATLAB we have used
meshgrid and surf to plot a
3 dimensional figure.**



Aluminium Gallium Arsenide



Here,

X axis represents E_{g1} ,

Y axis represents E_{g2} ,

Z axis represents P_{out}

Maximum Value of $P_{out} = 611.5 \text{ W/m}^2$ for

corresponding $E_{g1} = 2.168 \text{ eV}$ and $E_{g2} = 1.5 \text{ eV}$

$\text{Al}_x \text{Ga}_{1-x} \text{As}$

Aluminium Gallium Arsenide

Maximum Value of P_{out} was found in **Case 1**.

$$P_{out} = 640.1 \text{ W/m}^2$$

for

corresponding $E_{g1} = 2.168\text{eV}$ and $E_{g2} = 1.424\text{eV}$

From Bandgap Equations,

$$\text{If } E_g < 1.985, E_g = 1.424 + 1.247x$$

$$\text{If } E_g > 1.985, E_g = 1.9 + 0.125x + 0.143x^2$$

Here, $E_{g1} = 2.168\text{eV}$ which is greater than 1.985eV . It is the maximum value of E_g for the corresponding bandgap equation. Which can be obtained at mol fraction $x_1 = 1$.

$$\text{@}x_1=1, E_{g1}(\text{AlAs})= 2.168\text{eV}$$

Here, $E_{g2} = 1.424\text{eV}$ which is less than 1.985eV . It is the minimum value of E_g for the corresponding bandgap equation. Which can be obtained at mol fraction $x_2 = 0$.

$$\text{@}x_2=0, E_{g2}(\text{GaAs})= 1.424\text{eV}$$

Comparison



Gallium Arsenide Antimonide

Maximum Output Power,
 $P_{out} = 707.5 \text{ W/m}^2$

$$\begin{aligned} \text{Efficiency, } \eta &= P_{out}/P_{in} \times 100\% \\ &= 707.5/1200 \times 100\% \\ &= 58.9\% \end{aligned}$$



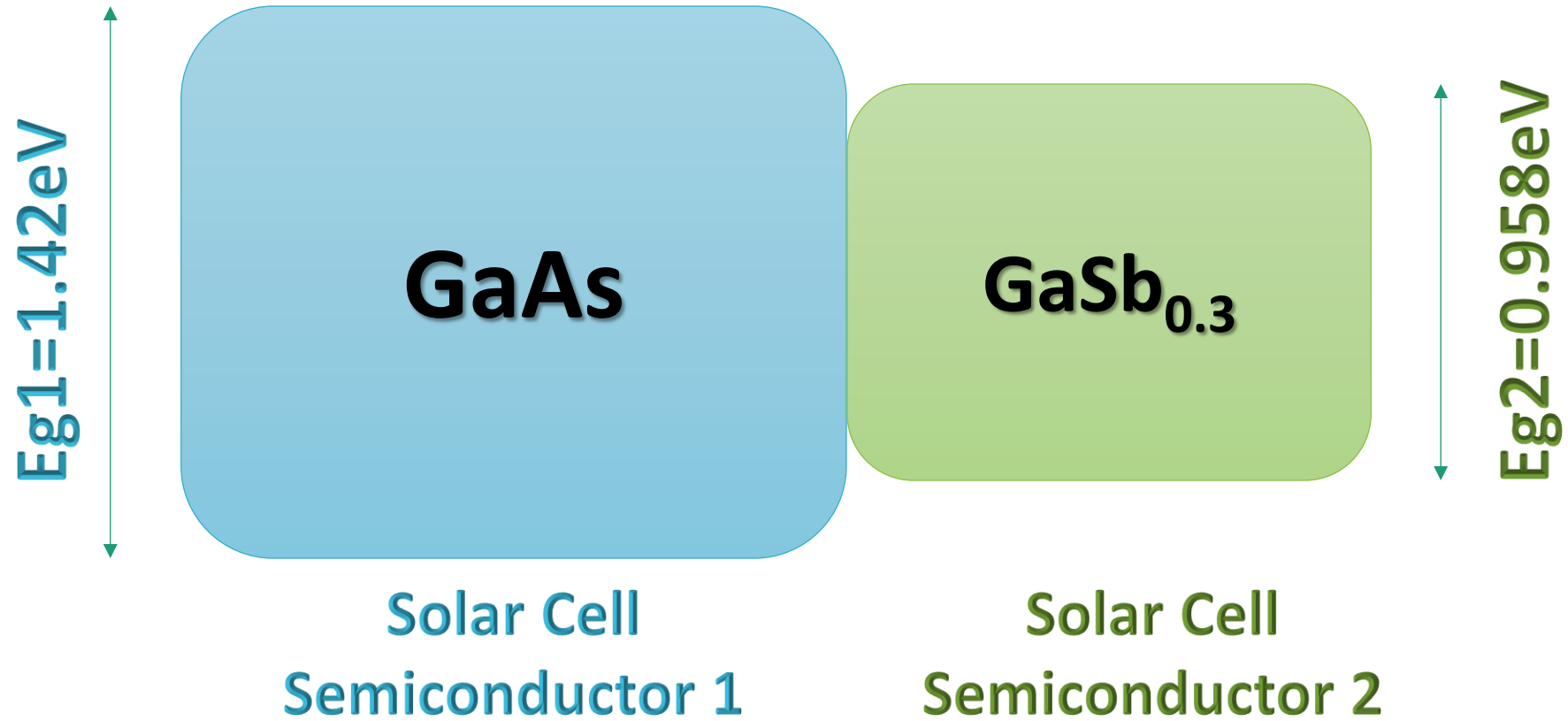
Aluminium Gallium Arsenide

Maximum Output Power,
 $P_{out} = 640.1 \text{ W/m}^2$

$$\begin{aligned} \text{Efficiency, } \eta &= P_{out}/P_{in} \times 100\% \\ &= 640.1/1200 \times 100\% \\ &= 53.3\% \end{aligned}$$

Among the Ternary Semiconductors $\text{GaAs}_{1-x}\text{Sb}_x$ is more efficient. So we should select $\text{GaAs}_{1-x}\text{Sb}_x$

Two Stage Tandem Solar Cell



Environmental Impact

**Disposal of the device made with $\text{GaAs}_{1-x}\text{Sb}_x$
can be harmful for the environment**

- **Disposed to water**
- **Disposed to earth**
- **Disposed by incineration**

Environmental Impact

Solar cell systems do not produce any toxicity. The health and environmental impacts are also associated with the fabrication process of solar cells and solar panels.

Material	Toxicity Type	Critical effects
Arsenic	Poison	Cancer, lung
Arsine	Highly toxic gas	Blood, kidney
Antimony	Poison	Skin disease, respiratory and immune systems
Hydrochloric acid	Corrosive material	Destroys human cells and tissues
Methane	Flammable gas	Fire hazard
Trimethyl gallium	Pyrophoric liquid	Fire hazard

Thank You